

Investigating Drug–Target Association and Dissociation Mechanisms Using Metadynamics-Based Algorithms

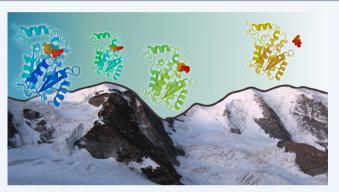
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CONSPECTUS: This Account highlights recent advances and discusses major challenges in the field of drug-target recognition, binding, and unbinding studied using metadynamics-based approaches, with particular emphasis on their role in structure-based design. Computational chemistry has significantly contributed to drug design and optimization in an extremely broad range of areas, including prediction of target druggability and drug likeness, de novo design, fragment screening, ligand docking, estimation of binding affinity, and modulation of ADMET (absorption, distribution, metabolism, excretion, toxicity) properties. Computationally driven drug discovery must continuously adapt to keep pace with the evolving knowledge of the factors that modulate the



pharmacological action of drugs. There is thus an urgent need for novel computational approaches that integrate the vast amount of complex information currently available for small (bio)organic compounds, biologically relevant targets and their complexes, while also accounting accurately for the thermodynamics and kinetics of drug-target association, the intrinsic dynamical behavior of biomolecular systems, and the complexity of protein-protein networks. Understanding the mechanism of drug binding to and unbinding from biological targets is fundamental for optimizing lead compounds and designing novel biologically active ones. One major challenge is the accurate description of the conformational complexity prior to and upon formation of drug-target complexes. Recently, enhanced sampling methods, including metadynamics and related approaches, have been successfully applied to investigate complex mechanisms of drugs binding to flexible targets. Metadynamics is a family of enhanced sampling techniques aimed at enhancing the rare events and reconstructing the underlying free energy landscape as a function of a set of order parameters, usually referred to as collective variables. Studies of drug binding mechanisms have predicted the most probable association and dissociation pathways and the related binding free energy profile. In addition, the availability of an efficient open-source implementation, running on cost-effective GPU (i.e., graphical processor unit) architectures, has considerably decreased the learning curve and the computational costs of the methods, and increased their adoption by the community. Here, we review the recent contributions of metadynamics and other enhanced sampling methods to the field of drug-target recognition and binding. We discuss how metadynamics has been used to search for transition states, to predict binding and unbinding paths, to treat conformational flexibility, and to compute free energy profiles. We highlight the importance of such predictions in drug discovery. Major challenges in the field and possible solutions will finally be discussed.

1. INTRODUCTION

Approximately 12 years and over 1 billion dollars are required to develop one new medicine from the earliest stages of discovery to approval. Only a tiny percentage of the compounds entering the drug discovery pipeline reach the market. High-throughput screening (HTS) is one of the primary pharmaceutical methods for identifying lead compounds, but it has a high false positive rate. Candidates are not routinely translated into new drugs. This leads to the delay or failure of drug discovery projects.¹ The reasons for this include unclear target biology, poor druglike properties, low potency or selectivity, lack of efficacy, unexpected toxicity, and so forth. Failures in the later stages are extremely costly. It is therefore of paramount importance to identify early on lead compounds that interact with the target in a biologically relevant mechanism without inducing adverse modes of action.

In recent years, more dynamic models of molecular recognition have superseded Emil Fischer's rigid lock-and-key binding paradigm. The two major paradigms are now the induced-fit and the conformational selection models.² It is now increasingly clear that both mechanisms contribute to molecular recognition, which requires a dynamic description over time scales spanning several orders of magnitude to be

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fully understood. This presents a fundamental theoretical and computational challenge.³

Molecular dynamics (MD) simulations have greatly contributed to understanding molecular recognition phenomena as a fully dynamical process. But they are limited by the time scales that can be routinely sampled. Very recently, with the introduction of special-purpose machines such as Anton,⁴ the porting of MD codes to GPUs,⁵ and the evolution of parallel codes,^{6,7} the time scales accessible by fully atomistic MD have increased enormously. Simulations lasting up to a few ms (corresponding to 10^{12} time steps) are now possible. However, druglike molecules with long residence times (more than an hour) are common, and their unbinding cannot be observed by conventional MD calculations even when specialized hardware is used. This well-recognized limitation of MD is not limited to ligand binding and has led to the development of many innovative algorithms to enhance the sampling of the high-infree-energy states and the rare events that allow the crossing of very high-free-energy barriers.8

Enhanced sampling methods speed up conformational sampling by various means. Umbrella sampling,⁹ replica exchange,¹⁰ metadynamics,¹¹ steered MD,¹² accelerated MD,¹³ milestoning,¹⁴ transition path sampling,¹⁵ and their many combinations and derivatives are among the most widely used methods to enhance conformational sampling. Of the enhanced sampling methods that fully explore the binding mechanism, metadynamics,¹¹ especially in the well-tempered formulation,¹⁶ has emerged as a powerful approach for accelerating rare events and computing multidimensional free energy surfaces (FESs). Metadynamics is a family of methods that flatten the FES as a function of a set of *collective variables* (CVs), by introducing a bias that prevents the system from being trapped in local free energy minima (Figure 1). It shares

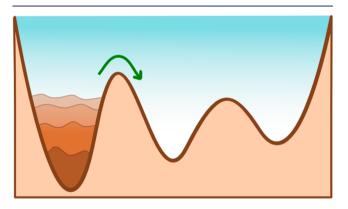


Figure 1. Pictorial representation of the way the metadynamics algorithm fills the free energy landscapes.

many similarities with other CV-based algorithms,^{17–22} and it has significantly evolved since its introduction. In contrast to other approaches, the reconstruction of the FES can be easily done in more than two dimensions and does not require an additional step (such as WHAM²³). It has been recently proved under general circumstances that well-tempered metadynamics provides an unbiased estimate of the free energy of the system projected onto predefined CVs, albeit the convergence time depends critically on how optimal are the CVs.²⁴ By "optimal CVs", one generally means a small set (2–4) of variables, function of atomic positions (e.g., distance), whose combination provide a good description of the reaction coordinate and is able to clearly distinguish between different free energy minima, kinetic basins, and binding pathways.

An open source plug-in implementation of metadynamics (PLUMED)²⁵ working with many widely used MD codes, as well as native implementations for NAMD, Desmond and other codes are available. In drug discovery, it can provide the location of cryptic pockets, an estimate of the binding free energy, including conformational changes of the target, as well as quantitative information on the metastable minima and transition states, which can be used to optimize the underlying ligand binding and unbinding kinetics.

In this Account, we review recent applications of metadynamics and compare them to other enhanced sampling approaches in drug-discovery-related endeavors, providing some perspective on the advantages, disadvantages, and practical applicability of these methodologies in academic and industrial settings.

2. PROTEIN-LIGAND BINDING AND UNBINDING

MD simulations have contributed in major ways to our understanding of molecular recognition and consequently to drug design. HIV-integrase inhibitors are a classic example. MD simulations revealed a previously unknown "trench", which was invisible to X-ray crystallography and traditional docking methods.²⁶ This suggested a way of designing stronger inhibitors binding to both the catalytic site and the newly discovered trench. A few notable examples in the literature deal with the long time span required for a binding process to occur spontaneously, by running MD simulations lasting a few microseconds.^{27–29} However, the time scale problem still hampers an extensive application of fully atomistic MD in drug design. In the context of accelerated sampling methods, metadynamics-based approaches have been successfully applied to drug design, particularly when major conformational motions play a role prior to and upon a drug binding to a target.

The first application of metadynamics to ligand-binding was carried out by Gervasio et al., who studied four different complexes.³⁰ The correct geometry corresponded to the deepest free energy minimum in all the cases, demonstrating the ability of the method to predict the correct binding mode. In the case of the trypsin-benzamidine complex, the algorithm was extensively compared to two-dimensional (2D) umbrella sampling and was found to provide the free energy reconstruction in less time and in good agreement with experimental data. In the case of the CDK2/staurosporine complex, albeit the correct docking pose was predicted, the computed binding free energy was significantly overestimated (20 vs 11 kcal/mol). This was due to a fast filling time (a total of only 8 ns), an inaccurate ligand force-field, the lack of a correction for the standard volume of the unbound state, and suboptimal CVs. Nowadays, thanks to the significant increase in available computational resources and the introduction of the well-tempered variant of metadynamics, the bias is added at a much slower rate and the FES is usually filled in hundreds if not thousands of nanoseconds. The ligand force field still remains a major issue. While there has been a significant progress in the accuracy of protein force fields,^{31,32} ligand force fields lag behind, and despite recent progress,³³ careful reparameteriza-tion of ligands based on ab initio calculations is strongly advised. The need to choose a set of optimal CVs is the main drawback of metadynamics and other CV-based methods, despite major progress has been achieved in the design of

Accounts of Chemical Research

optimal coordinates. In the work of Gervasio et al.,³⁰ the choice of geometry-based CVs (distance and angle) was guided by three requirements: (i) a general applicability in molecular recognition, (ii) the need to distinguish all the minima along the reaction coordinate, and (iii) the need to keep the number of CVs low. The latter requirement is due to the fact that the time needed to "fill the minima" (see Figure 1) in principle increases exponentially with the number of the CVs.

In a subsequent study, Masetti et al. systematically explored different combinations of CVs and faster approaches to converge the free energy estimation.³⁴ A combination of docking, cluster analysis,³⁵ and metadynamics limited to exploring the internal cavity up to the transition state (*coarse-metadynamics*) improved scoring in conventional structure-based endeavors. To date, *coarse-metadynamics* together with optimal CVs is still one of the best ways to use metadynamics to rank different ligands binding to the same target. However, to compute the relative free energy of a set of similar compounds (having a similar binding pose), end-point methods, that is, methods that compute the difference of free energy between the bound and the unbound state, such as thermodynamic integration (TI) are in most cases more efficient than methods that compute the free energy along a physical association pathway.

Another early application was Branduardi et al.'s study of the binding mechanism of the tetramethylammonium (TMA) ion into human acetylcholinesterase (AChE).³⁶ The AChE peripheral anionic site (PAS), located at the entrance of the gorge, is solvent-exposed; therefore, plain MD simulations are able to identify stable docking conformations.³⁷ However, enhanced sampling approaches are needed to explore binding into the deep internal gorge.³⁸ Metadynamics was run with two CVs, the distance between ligand and the binding pocket, along with a new CV describing the cation– π interaction, leading to a well-converged free energy reconstruction. Cation– π interactions were shown to play a fundamental role in both the initial recognition between TMA and the PAS, and the gorge binding mechanism. Notably, the cation– π CV must be included to obtain a converged FES.

Beside protein–ligand systems, metadynamics has also been applied to study the association mechanism of molecules binding to DNA and RNA, for example, the interaction of ligands with DNA G-quadruplexes,³⁹ and the unbinding of the anticancer drug distamycin from the DNA minor groove.⁴⁰

Other methods, such as steered MD, have also been successfully used to characterize binding poses and to drive the selection of biologically active compounds. Colizzi et al. have exploited docking and steered MD to discern between active and inactive antimalarial ligands.⁴¹ By modeling the force that is required to pulling inhibitors out of their binding pocket, it was possible to separate strong binders from weak ones. Strongly bound inhibitors, which gave a flat profile (Figure 2). Compared to metadynamics, in steered MD the choice of the CV is less crucial, allowing a quick exploration of many unbinding events. However, steered MD is less efficient in computing fully converged free energy profiles.

3. RECONSTRUCTING THE FREE ENERGY LANDSCAPES

As discussed above, selecting a limited number of optimal CVs is a crucial step in metadynamics simulations to obtain a converged free energy landscape. The most commonly used

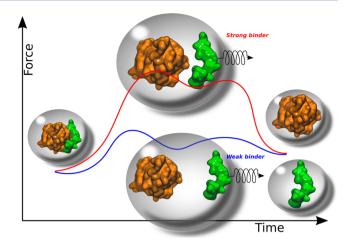


Figure 2. Force applied to the ligand by the spring calculated as a function of time, generating a force profile that reflects the strength of protein–ligand binding as reported in Colizzi et al.⁴¹

CVs in drug-binding simulations are geometry-based and involve a combination of distances, contacts, angles, and dihedrals. In complex cases, several attempts might be needed to find a proper combination of variables (suboptimal sets of variables are easily recognized as the reconstructed FES shows hysteresis as a function of time). The selection of effective CVs in ligand-(un)binding is further complicated by the need to fully explore (fill) the large conformational spaces available to the solvated ligand in the bulk, once unbound from its target.

Metadynamics is not efficient in exploring relatively flat, diffusive energy landscapes and its use to explore unbound and weakly bound states is not recommended. In such cases, other methods such as swarms of free MD trajectories and Markov-state models²⁸ or some metadynamics variants such as Reconnaissance metadynamics or bias-exchange are more effective. When the binding area is known, this issue can be solved by the use of restraints on the explored area or optimal CVs such as the path collective variables (PCVs).⁴²

In analogy to similar strategies used with umbrellasampling,⁴³ various restraining potentials have been used. An interesting choice is to use a reverse *funnel-shaped* potential that limits the space available to the ligand once it has undocked.⁴⁴ *Funnel* metadynamics was used to revisit some of the systems previously studied with geometrical CVs or PCVs, including the trypsin—benzamidine complex and the binding of the drug SC-558 to cyclooxygenase 2 (COX-2, see below). Thanks to the restraining potential, the unbound state can be thoroughly sampled leading to a faster convergence of the energy landscape.

A perhaps more elegant solution is the use of PCVs that describe the position of the system in the configuration space relative to an optimal exit path, in terms of progression along the path and distance from the path itself. PCVs are able to describe complex association pathways including target conformational changes and to restrain the sampling of the unbound state, improving convergence considerably. In the past few years, the PCV approach has been successfully used to study complex conformational changes and drug-target binding.^{45,46} Fidelak et al.⁴⁷ used PCVs to study the binding of a congeneric series of ligands to CDK2. Identifying the correct order of activity of the compounds was nontrivial, due to the lack of interaction between the protein and the aryl-pyrimidine substituents differentiating the homologous inhib-

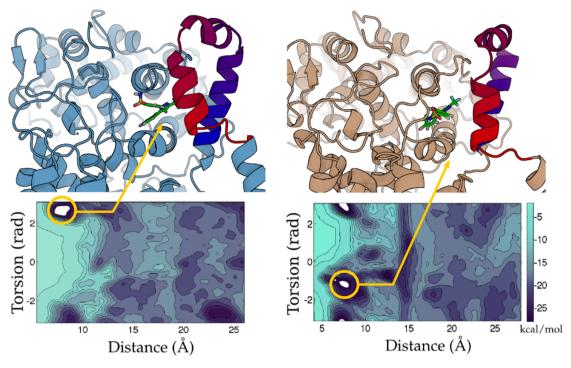


Figure 3. Free energy surfaces associated with the binding of the SC558 inhibitor to COX-1 and COX-2 as a function of the distance between the ligand and the cavity and the relative orientation of SC558.

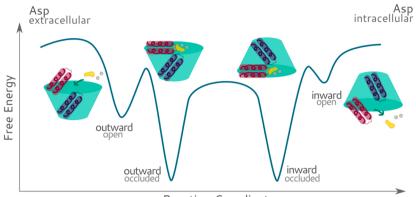
itors. Using well-tempered metadynamics and PCV, the authors converged a full association free energy profile in remarkable agreement with the experimental binding energies. They also fully characterized the association mechanism. In a subsequent work, Saladino et al.⁴⁸ studied a well-known series of p38 inhibitors, frequently used to assess the performances of scoring algorithms, whose activity cannot be easily guessed. The authors moved a step further toward making metadynamics suitable for a pharmaceutical workflow and used a semiautomatic setup. With a similar intent, Mason and co-workers⁴⁹ at Heptares Therapeutics designed the MetaScore routine, a fast metadynamics-based protocol, used to predict the effect of different mutations on the binding of antagonists to the adenosine A2A receptor. The MetaScore scoring function is based on a two-step semiautomatic protocol that uses several adiabatic biased MD simulations to obtain an initial association path and construct PCVs, followed by short well-tempered metadynamics runs to calculate the full binding free energy profile. Using this approach, the authors predicted the correct binding energy changes resulting from active site mutations. PCVs were also used by Provasi et al.⁵⁰ to access the free energy of binding of several ligands to another G-proteincoupled receptor, the β 2-adrenergic receptor.

When the binding is accompanied by large-scale conformational changes, the choice of CVs can be daunting. In these cases, Parallel Tempering Metadynamics (PT-MetaD),⁵¹ an approach that couples metadynamics with a replica exchange algorithm, can enhance the sampling of hidden (slow) degrees of freedom and greatly improve convergence. It has been used to compute the FES associated with the inactive-to-active conformational changes of the cancer target epidermal growth factor receptor.⁴⁶ PT-MetaD, albeit being a very expensive approach, is the method of choice when large-scale conformational changes are involved and a converged FES is required. Finally, the number of CVs is not an issue in Bias Exchange Metadynamics (BEMD)⁵² and Reconnaissance Metadynamics.⁵³ The latter uses a self-learning algorithm that clusters the visited conformations on-the-fly and bias along a onedimensional clustering-based CV. These approaches have been effectively used to explore very complex molecular recognition events and are very beneficial when little information is known about the association mechanism and the location of the binding cavity. Both methods, however, require extra efforts to reconstruct the free energy landscapes, and thus are best used as a means to explore different binding mechanisms.

4. PHARMACEUTICAL APPLICATIONS

Metadynamics was recently used to understand the differences in the binding mechanisms of a nonsteroidal anti-inflammatory ligand (SC-558) to COX-1 and COX-2 isoforms.⁵⁴ Selective inhibition of COX isoforms has been actively sought, leading to new generations of widely used COX-2 selective drugs. Recently, significant effort has been made to understand the selectivity and the differences in the binding mechanism of inhibitors to COX-1 and COX-2. In this study, metadynamics and the PCV method⁴² were used to describe the (un)binding of a ligand and the associated conformational change of three α -helices that form a narrow gate through which the ligand exits from the protein. In the case of the COX-2 isoform, two separate and almost equally favorable minima in the FES are found (Figure 3), while only one deep minimum was found in COX-1. The existence of two binding modes in COX-2 only explains both the selectivity of the ligand and, interestingly, the increased residence time of the ligand in COX-2 relative to COX-1.

A similar approach was used to study the aspartate uptake and internal release mechanism through a membrane protein, namely, the amino acid transporter from *Pyrococcus horikoshii*.⁵⁵ The transport of the substrate is assisted by the cotransport of



Reaction Coordinate

Figure 4. Schematic representation of the free energy landscape of substrate uptake and release by the glutamate transporter. Adapted with permission from ref 55. Copyright 2012 American Chemical Society.

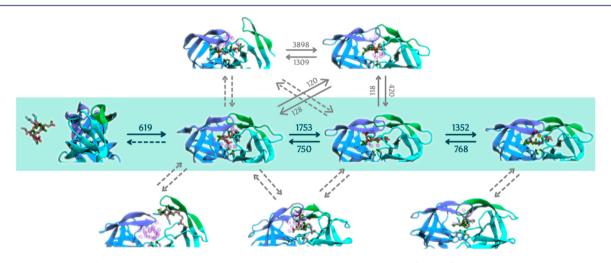


Figure 5. Kinetic model of the substrate binding to HIV-1 protease. Adapted with permission from ref 59. Copyright 2009 American Chemical Society.

Na⁺ ions and is mediated by the concerted motion of the transmembrane domain and the two hairpins (HP1 and HP2) that regulate access to either side of the membrane. In addition, it is known that this transporter may assume at least four different conformational states, which further challenges the study of the substrate translocation across the channel. Grazioso et al. used PCVs to enhance the highly cooperative transition upon substrate binding and transport across the membrane.⁵⁵ The authors thus described the complete opening mechanism, and the associated free energy profile, involving a large-scale motion of HP2, which allowed substrate uptake, with a similar role played by HP1 in the inward release mechanism (Figure 4).

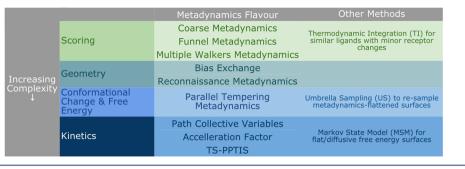
As a final case study, PT-metaD with PCVs and BEMD have been used to study the mode of action of a novel inhibitor of the fibroblast growth factor receptor (FGFR).⁵⁶ FGFRs are tyrosine kinase receptors whose signaling pathway is involved in cell growth, proliferation, and survival. Due to their key role in angiogenesis, FGFR deregulation can lead to the development of malignancies, so this family of receptors is a wellstudied target for cancer treatment. The first extracellular FGFR inhibitor, SSR128129E (hereafter referred to as SSR), was discovered by the same authors through a high-throughput screening campaign. However, several experimental techniques, ranging from NMR to crystallography, failed to identify the binding site of SSR, due to the inherent flexibility of the extracellular domain of FGFR, comprising three immunoglobulin-like domains. Using BEMD, the authors observed the elongation of a small α -helix in the D3 domain in the presence of the drug, leading to the opening of a pocket not present in the apo structure of FGFR. Metadynamics simulations with PCV correctly predicted the dissociation constant of the compound and of several lower potency analogues, validating the in-silico-derived mode of action. As a further proof, the authors used the observed binding mode as a template for rationally designing new inhibitors, identifying 16 new compounds with varying degrees of activity. Once again, the computationally predicted free energy was in remarkable agreement with experimental results.

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5. BINDING AND UNBINDING KINETICS ESTIMATION

Obtaining reliable estimates of the kinetics constants associated with ligand (un)binding is an ambitious goal of modern computational drug discovery, with significant potential benefits in the lead discovery and the lead optimization steps of the drug discovery process.⁵⁷ The possibility of reconstructing and estimating the binding free energy along physical association and dissociation pathways provides quantitative information on the metastable minima and transition states. These can then be used to optimize the underlying kinetics to achieve therapeutically safe and differentiated responses. Researchers increasingly recognize the importance of ligand target residence time in fine-

Table 1. Summary of the Suggested Approaches for Different Drug Discovery Applications



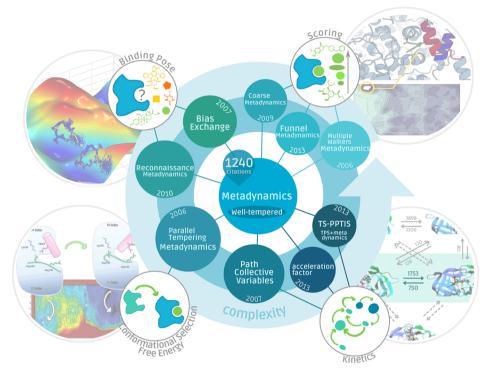


Figure 6. Schematic representation of the use of metadynamics in drug design, from scoring and binding pose prediction to thermodynamics and kinetics profiles.

tuning in vivo efficacy and toxicology.⁵⁸ Knowledge of the metastable states and the residues involved in a ligand's access to the binding cavity could lead to the design of novel drugs that are more potent and selective for therapeutically relevant targets.

Metadynamics in the bias-exchange variant was used by Pietrucci et al. to investigate the key events in the binding of the p2-NC peptide to the human immunodeficiency virus type-1 protease (HIV1-1 PR).⁵⁹ The free energy was computed as a function of seven CVs, in order to characterize the physical interactions that might play a role in stabilizing the complex and the transition states. From the BEMD trajectories, a thermodynamic and kinetic model of the binding process was constructed based on the weighted-histogram approach (Figure 5). Remarkably, mutation of residues along the access region identified by BEMD caused resistance to some United States Food and Drug Administration approved peptidomimetic drugs, providing an experimental validation. Individual water molecules at the interface between ligand and enzyme played a pivotal role throughout the binding process, and a CV describing "interfacial water" was needed to discriminate different intermediate states.

In the absence of large barriers to binding/unbinding, Markov state models (MSM)⁶⁰ might be more efficient than metadynamics-based algorithms in reconstructing the binding kinetics. A collection of 495 100 ns long trajectories was used to reconstruct the full binding kinetics of the trypsin/benzamidine complex, revealing the existence of long-lived binding intermediates.

Recently, Tiwary and Parrinello⁶¹ adapted to metadynamics the concept of "acceleration factor" introduced by Grubmüller¹⁷ for Conformational Flooding. Provided that no bias is deposited in the transition state region, this approach can predict the rate of the transition, with minimal additional computational cost. Albeit the authors applied their approach to a toy model, preliminary data indicates that it can be applied to predict the binding kinetics in simple (two-state) systems. A more general (and more computationally expensive) approach was developed by Juraszek and co-workers.⁶² By combining PT-MetaD-derived free energies with a transition path sampling (TPS) approach, they devised a robust method that can be applied to complex ligand-binding events with multiple intermediate states and diffusive barriers. Indeed, the *Transition State-Partial Path Transition Interface Sampling* was explicitly

Accounts of Chemical Research

devised to overcome the difficulties of available methods in predicting the kinetics rates of drug binding phenomena. Introducing a new recursive expression for the transmission coefficient, the authors designed a robust approach with very general applicability and high accuracy.

6. PERSPECTIVES

In Table 1 and Figure 6, we summarize the suggested metadynamics-based approach to use in drug discovery. In particular, we address the cases of cryptic binding sites, scoring different ligands binding to the same target, full free energy reconstruction with or without large conformational changes, and (un)binding kinetics. As we have detailed in the preceding sections, metadynamics-based approaches are not always the most effective choice. In Figure 7, we report on the major



Figure 7. SWOT (Strengths, Weakness, Opportunities, and Threats) analysis for metadynamics.

strengths and weaknesses of these methods. One major drawback of metadynamics is still related to the choice of CVs, which can compromise a proper reconstruction of the FES associated with the process under investigation. In this scenario, some recent and ongoing developments, particularly suited to drug discovery, are the use of nonlinear dimensionality reduction techniques to automatically find the optimal CVs, new CVs to find and score cryptic binding sites, and, going beyond the current methods, a new variational approach to construct an optimal bias potential and reconstruct the FES.⁶³ All these evolutions will contribute to a further expansion of metadynamics approaches to drug discovery, widening the plethora of computational tools utilized to estimate binding and unbinding free energy and kinetics, and eventually to accelerate the discovery of novel bioactive compounds.

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The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

MD, molecular dynamics; CV, collective variable; FES, free energy surface; PCV, path collective variables; PT-MetaD, parallel tempering metadynamics; BEMD, bias-exchange molecular dynamics; MSM, Markov state model

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